Reactive Dye

Reactive dye is the dye that can react with a fibre to **form a covalent link**, that is forming a **permanent attachment** in the fibre and could not be **removed by repeated treatment** with boiling water under **neutral conditions**.

Or,

A dye, which is **capable of reacting chemically** with a substrate to **form a covalent** substrate linkage, is known as reactive dyes.

Properties or Characteristics of reactive dye:

- ✓ Reactive dyes are anionic dyes,
- ✓ use for dyeing cellulose, protein and polyamide fibres.
- ✓ Reactive dyes are found in **power**, **liquid and print paste** form.
- ✓ Contains reactive group in its structure
- ✓ During dyeing the reactive group of this **dye forms covalent bond with fibre polymer**
- ✓ Reactive dyes are **soluble in water**.
- ✓ They have **very good light fastness with rating about 6**.
- ✓ Reactive dyes have **very good wash fastness with rating about 4-5** due to **strong covalent bonds** formed between fibre polymer and reactive group of dye.
- ✓ Reactive dye gives brighter shades and has moderate rubbing fastness.

- ✓ Dyeing **method of reactive dyes is easy**. It requires less time and low temperature for dyeing.
- ✓ Reactive dyes are **comparatively cheap**
- ✓ Reactive dyes have good **perspiration fastness with rating 4-5**.

Structure of Reactive Dye

Reaction:

 $D\text{-SO}_2\text{-CH}_2\text{-CH}_2\text{-OSO}_3\text{Na} + \text{OH-cell} \quad \rightarrow \quad D\text{-SO}_2\text{-CH}_2\text{-CH}_2\text{-O-cell} + \text{NaHSO}_3$ Here,

D= dye part.

Cell = cellulosic polymer.

Why reactive dye is so called?

Reactive dyes are so called because this is the **only type of dye**, which has **reactive group**, and that reactive group reacts **chemically with fibre polymer molecules** and **form covalent bond**. This covalent bond is formed between the reactive group and terminal –OH (Hydroxyl) group of cellulosic fibre and wool fiber or between reactive group and terminal -NH2 (Amino) group of polyamide polymer. The strength of this covalent bond is more than ionic bond, hydrogen bond and Vander Waal's force of attraction.

For this reason, the dyes are so called. They are also called 'fibre reactive group'.

History:

On the occasion of 100 year's celebration of synthetic dyes manufacturing, two chemists of ICI company (UK) named Stephen and Rattee tried to manufacture a new dyestuff. Thus, they succeed to invent a new dye in 1965, which was named REACTIVE DYE. This was manufacture for dyeing cellulosic fabrics. The first three reactive dyes were PROCION YELLOWR, PROCION BRILLIANT RED 2B and PROCION BLUE 3G for this effort they were awarded gold medal of the society of dyes and colorists.

Trade names of reactive dye:

Trade name	Manufacturer	Country
Procion	I.C.I	U.K
Ciba cron	Ciba	Switzerland
Remazol	Hoechst	Germany
Levafix	Bayer	Germany
Reactone	Geigy	Switzerland
Primazin	BASF	Germany
Drimarine	Sandoz	Switzerland

Popularity of reactive dye:

Reactive dyes are mostly used for **dyeing cellulosic fibres**. At past **cellulosic fibres** were **dyed with direct and vat dyes**, but after the introduction of reactive dyes, **their utility has become limited**. Reactive dyes are superior to direct dye in the following aspects:

- ➤ Ability to procedure bright shades of wide range.
- ➤ High leveling quality.
- ➤ Good washing fastness.
- > Good light fastness.

And it is superior to vat dyes in the following aspects:

- > Simple dyeing method therefore one stage dyeing.
- ➤ Low temperature dyeing (below 100°C)
- ➤ Lower cost, i.e. cheaper.

Again its **dyeing process is fast** and gives **brighter shades than metallized azo dyes**. For the above reasons reactive dyes are more popular.

Classification of reactive dyes:

Reactive dyes may be classified in various ways as below:

A. On the basis of reactive group:

i. Halogen derivatives

- ✓ Triazine group
- ✓ Pyridine group
- ✓ Quinoxaline dyes

Example:

- ✓ Triazine derivatives: procion, cibacron.
- ✓ Pyridine derivatives: reactone
- ✓ Quinoxaline derivatives: levafix.

ii. Activated vinyl compound:

- ✓ Vinyl sulphone
- ✓ Vinyl acrylamide
- ✓ Vinyl sulphonamide.

Example:

- ✓ Vinyl sulphone: remazol
- ✓ Vinyl acrylamide: primazine
- ✓ Vinyl sulphonamide: levafix.

B. On the basis of reactivity:

- ✓ **Lower reactive dye:** Here pH is maintained 12-12.5 by using NaOH in bath.
- ✓ **Medium reactive dye:** here pH is maintained 11-12 by using Na₂CO₃ in dye bath.
- ✓ **Higher reactive dye:** here pH is maintained 10-11 by using NaHCO₃ in dye bath.

C. On the basis of dyeing temperature:

a. Cold brand:

These types of dyes contain reactive group of high reactivity. So dyeing can be done in lower temperature i.e. $30^{\circ}-60^{\circ}$ C.

For example: PROCION M, LIVAFIX E.

b. Medium brand:

This type of dyes contains reactive groups of moderate reactivity. So dyeing is done in higher temperature than that of cold brand dyes i.e. in between 60°-70°C temperatures.

For example, Remazol, Livafix are medium brand dyes.

c. Hot brand:

This type of dye contains reactive groups of least reactivity. So high temperature is required for dyeing i.e. 70°-95°C temperature is required for dyeing.

For example PRICION H, CIBACRON are hot brand dyes.